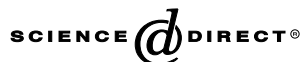




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Optical Materials xxx (2005) xxx–xxx



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## Luminescence of erbium-doped bismuth–borate glasses

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Received 26 May 2005; accepted 12 July 2005

### Abstract

Absorption and luminescence properties of erbium ions in the binary glass system bismuth oxide ( $\text{Bi}_2\text{O}_3$ )–boric oxide ( $\text{B}_2\text{O}_3$ ) are measured for the composition range 25–65 mol%  $\text{Bi}_2\text{O}_3$ . A Judd–Ofelt analysis of the typical erbium bands in the absorption spectra reveals comparably high Judd–Ofelt coefficients. This indicates a substantial mixing of other electronic configuration into the  $4f^N$  configuration by the random crystal fields in the glasses. All coefficients are decreasing with increasing  $\text{Bi}_2\text{O}_3$  content, this effect being most pronounced with  $\Omega_2$ . Luminescence decay times and radiative efficiencies show an expressed dependence on the glass composition. Radiative efficiencies of all luminescence bands increase with increasing  $\text{Bi}_2\text{O}_3$  content—accompanied, however, by a slight narrowing of the bands. Except the common luminescence bands of erbium, upconversion luminescence at a wavelength of 0.54  $\mu\text{m}$  could be detected.

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PACS: 42.70.Ce; 78.55.Qr

Keywords: Erbium doping; Bismuth–borate glass; Luminescence

### 1. Introduction

Since about 40 years, rare earth ions like  $\text{Er}^{3+}$  have attracted much interest as luminescence centers in crystals as well as in glasses [1–6]. Presently, special attention is paid to erbium due to its emission band at 1.53  $\mu\text{m}$  which makes it an ideal element for applications in the field of optical data transmission (erbium-doped fiber amplifier). Luminescence properties of  $\text{Er}^{3+}$  have been studied in crystalline hosts including garnets and related materials [7–11] where the suitability for lasers in the near infrared range has been demonstrated. Glassy host materials investigated include silicate glasses [12,13], germanate glasses [14], phosphate glasses [15,16], sul-

phide glasses [17,18], tellurite glasses [19] as well as more complicated glass systems like  $\text{Na}_2\text{O} \cdot \text{Ca}_3\text{Al}_2\text{Ge}_3\text{O}_{12}$  [20] or ZBLAN ( $\text{ZrF}_4 \cdot \text{BaF}_2 \cdot \text{LaF}_3 \cdot \text{AlF}_3 \cdot \text{NaF}$ ) [21], probably better suited for fiber amplifiers.

In glasses, various relevant properties—as e.g., the refractive index—can be tuned by the addition of heavy ions like lead or bismuth. Thus systems consisting of a light and a heavy component, preferably allowing for a broad composition range, are most suitable for applications where a wide tunability of the properties is desired. One of these systems is the binary phase boric oxide ( $\text{B}_2\text{O}_3$ )–bismuth oxide ( $\text{Bi}_2\text{O}_3$ ) [22]. Glasses can be prepared in this system with  $\text{Bi}_2\text{O}_3$  contents ranging from about 25–65 mol%. Several authors have recently studied the properties of bismuth borate glasses, undoped [23],  $\text{Er}^{3+}$ -doped [24] or  $\text{Nd}^{3+}$ -doped [25], however, in a restricted composition range. Becker could show, that the refractive index can be widely

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51 varied [23], Chen et al. proved that high luminescence  
52 efficiencies are possible—especially in  $\text{Nd}^{3+}$ -doped glass  
53 [25]. Tanabe et al. [26] showed for  $\text{Er}^{3+}$  doped  $\text{Bi}_2\text{O}_3 \cdot$   
54  $\text{B}_2\text{O}_3 \cdot \text{SiO}_2$  glasses that the luminescence lifetime and  
55 efficiency decrease with increasing  $\text{B}_2\text{O}_3$  content. Similar  
56 results were found by Yang et al. [27] for  $\text{Bi}_2\text{O}_3 \cdot \text{B}_2\text{O}_3 \cdot$   
57  $\text{Na}_2\text{O}$  glasses.

58 In this work, we present a comprehensive analysis of  
59 the optical properties—especially the luminescence  
60 properties—of  $\text{Er}^{3+}$  doped bismuth borate glasses for  
61 the complete composition range spanning from 25 to  
62 65 mol% bismuth oxide. Besides the refractive indices  
63 and the absorption properties, the influence of the com-  
64 position on the luminescence linewidth and the lumines-  
65 cence efficiency is studied.

## 66 2. Experimental

67 Samples of  $\text{Er}^{3+}$  doped bismuth borate glasses  
68  $(\text{Er}_2\text{O}_3)_y(\text{Bi}_2\text{O}_3)_x(\text{B}_2\text{O}_3)_{1-x-y}$  with compositions span-  
69 ning the range  $x = 0.25 \dots 0.65$  and  $y = 0.0025$  were  
70 fabricated from melts of corresponding compositions.  
71 The melts were thoroughly homogenized at a tempera-  
72 ture of  $\approx 1000$  K and then cooled down to room tempera-  
73 ture at a rate of 100 K per hour. To study the influence  
74 of higher  $\text{Er}^{3+}$  doping, in addition, samples with  $y =$   
75  $0.0075$  and  $y = 0.025$  were prepared for  $x = 0.25$ .

76 For the absorption measurements, a Bruins Omega  
77 10 spectrometer was used working in the near ultravio-  
78 let, visible, and near infrared spectral region. The  
79 absorption data were measured on plate-shaped samples  
80 with a thickness of approximately 1 mm. The data were  
81 corrected for reflection using previously measured  
82 refractive index values of the same samples [28]. For  
83 the correction, the Sellmeier fit formula provided there  
84 for a wavelength range from 0.4 to 1.53  $\mu\text{m}$  was applied.  
85 From this fit formula, also all refractive index data nec-  
86 essary for calculations were derived.

87 Luminescence measurements were performed in  
88 reflection geometry on platelet samples using several  
89 setups. For the spectra at 1.53  $\mu\text{m}$ , a 1.48  $\mu\text{m}$  or a  
90 0.8  $\mu\text{m}$  laser diode was used for excitation, an Amko  
91 monochromator and a suitable photodiode for detec-  
92 tion. For all other spectra, a 0.8  $\mu\text{m}$  semiconductor  
93 laser working in CW mode was applied as excitation  
94 source. The detection system consisted of a Jobin-Yvon  
95 TRIAX 180 spectrometer combined with either an  
96 infrared, red or green sensitive photomultiplier,  
97 depending on the spectral region of the respective mea-  
98 surement. For the decay measurements the samples  
99 were excited by a 0.8  $\mu\text{m}$  or a 1.48  $\mu\text{m}$  semiconductor  
100 laser in pulsed mode, appropriate photodiodes and  
101 photomultipliers rendered the time dependence of the  
102 luminescence.

## 3. Results and discussion

### 3.1. Absorption spectra and Judd–Ofelt analysis

105 For the short wavelength absorption edge of un-  
106 doped bismuth borate glasses a strongly expressed  
107 dependence on the  $\text{Bi}_2\text{O}_3$  content had been found previ-  
108 ously [28]. Doped glasses show a similar behavior for the  
109 absorption edge. The typical erbium absorption bands,  
110 however, exhibit only a very slight yet expressed depen-  
111 dence on the glass composition. This is shown in Fig. 1  
112 for the two compositions  $x = 0.25$  and  $x = 0.65$  (erbium  
113 content  $y = 0.0025$ ).

114 Electric dipole transitions between two states of  $4f^N$   
115 configuration of rare earth ions, which are forbidden  
116 for free ions, become allowed in the crystal field by mix-  
117 ing into the  $4f^N$  configuration another configuration  
118 ( $4f^{N-1}5d^1$ ) of opposite parity. The matrix elements of  
119 the electric dipole operator are calculated by considering  
120 the crystal field as a first-order perturbation.

121 According to the Judd–Ofelt theory [29,30], the effect  
122 of the crystal field can be described by three intensity  
123 parameters  $\Omega_{2,4,6}$ . The line strength for the electric di-  
124 pole transition  $S^{\text{ed}}$  then is given by

$$S^{\text{ed}}[(S, L)J; (S', L')J'] = \sum_{t=2,4,6} \Omega_t |\langle (S, L)J || U^{(t)} || (S', L')J' \rangle|^2, \quad (1)$$

125 where  $\Omega_t$  are the Judd–Ofelt parameters,  $U^{(t)}$  are the unit  
126 tensor operators, and  $(S, L)J$  and  $(S', L')J'$  represent the  
127 initial and final states of the transition.

128 For magnetic dipole transitions, the line strength  $S^{\text{md}}$   
129 is defined as

$$S^{\text{md}}[(S, L)J; (S', L')J'] = \frac{1}{4m^2c^2} |\langle (S, L)J || L + 2S || (S', L')J' \rangle|^2 \quad (2)$$

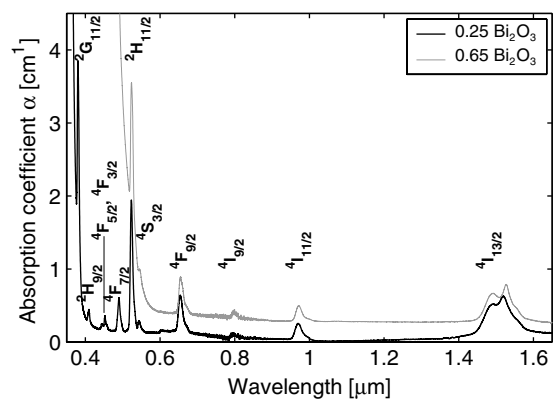


Fig. 1. Absorption spectra of  $\text{Er}^{3+}$  doped bismuth borate glasses for two glass compositions ( $x = 0.25$  and  $x = 0.65$ , for better visibility the spectrum for  $x = 0.65$  is shifted upwards). The absorption bands correspond to transitions from the erbium ground state  $4I_{15/2}$  to the excited states assigned.

136 with  $m$ —electron mass and  $c$ —velocity of light. Numerical  
137 values for the matrix elements in Eqs. (1) and (2) can  
138 be calculated according to Carnall et al. [31] and Weber  
139 [7]. The integral intensity over an absorption band then  
140 can be written as

$$141 \int_{\text{band}} \alpha d\lambda$$

$$143 = \frac{2\pi^2 e^2 \bar{\lambda} \rho}{3\epsilon_0 c h (2J+1)n^2} \cdot \left( \frac{n(n^2+2)^2}{9} S^{\text{ed}} + n^3 S^{\text{md}} \right), \quad (3)$$

144 where  $\alpha$  is the absorption coefficient,  $\bar{\lambda}$  the mean absorp-  
145 tion wavelength,  $e$  the electron charge,  $\rho$  the erbium con-  
146 centration,  $\epsilon_0$  the vacuum permittivity,  $h$  Planck's  
147 constant,  $(2J+1)$  the multiplicity of the ground level,  
148 and  $n$  the refractive index.

149 Using Eq. (3) the Judd–Ofelt coefficients were calcu-  
150 lated applying a least-squares fit procedure to the inte-  
151 gral absorption intensities of the various bands in the  
152 measured absorption spectra. The results are listed in  
153 Table 1.

154 All Judd–Ofelt coefficients exhibit comparably high  
155 values. While  $\Omega_4$  is approximately constant,  $\Omega_2$  and  $\Omega_6$   
156 are decreasing with increasing  $\text{Bi}_2\text{O}_3$  content. This indi-  
157 cates that short range structural disorder effects—ex-  
158 pressed by  $\Omega_2$ —are decreasing, whereas covalency  
159 effects in the Er–O bonds—showing up in  $\Omega_6$ —are  
160 increasing with increasing  $\text{Bi}_2\text{O}_3$  content. The values  
161 for the Judd–Ofelt coefficients previously given by Chen  
162 et al. [24] for compositions between 0.25 and 0.45 differ  
163 slightly from our values but show the same trends. The  
164 deviations may be attributed to differences in the reflec-  
165 tion corrections applied to the absorption measure-  
166 ments. For these corrections, we used the wavelength  
167 dependent Sellmeier description of the refractive indices  
168 [28] whereas Chen et al. did not apply such a correction  
169 to their measurements.

### 170 3.2. Luminescence at 1.53 $\mu\text{m}$

171 For applications in optical communication systems  
172 most important is the emission band at 1.53  $\mu\text{m}$  arising  
173 from the transition  $^4\text{I}_{13/2} \rightarrow ^4\text{I}_{15/2}$ . This emission can be  
174 pumped through the corresponding absorption transi-

Table 1

Calculated Judd–Ofelt parameters  $\Omega_i$  (in units of  $10^{-24} \text{m}^2$ ) as a function of the glass composition ( $\text{Bi}_2\text{O}_3$  content) for an  $\text{Er}_2\text{O}_3$  content of  $y = 0.0025$

$\text{Bi}_2\text{O}_3$ content	$\Omega_2$	$\Omega_4$	$\Omega_6$
0.25	4.58	1.44	1.53
0.35	3.72	1.16	1.15
0.45	3.56	1.11	1.29
0.55	3.45	1.18	1.05
0.65	2.56	1.23	0.82

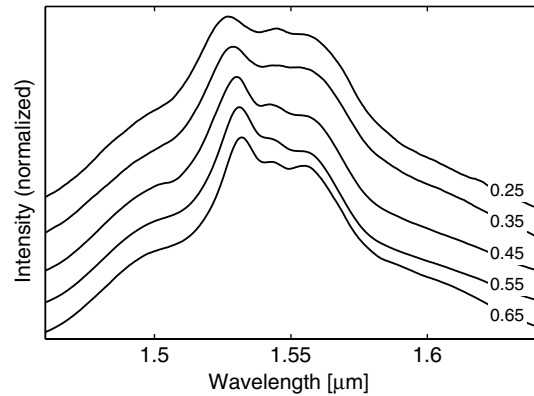


Fig. 2. 1.53  $\mu\text{m}$  emission spectra of  $\text{Er}^{3+}$  ions in bismuth borate glass samples of different compositions ( $\text{Bi}_2\text{O}_3$  content indicated at the right side of the spectra). Pump wavelength was 0.8  $\mu\text{m}$ .

tion  $^4\text{I}_{15/2} \rightarrow ^4\text{I}_{13/2}$  using an appropriate semiconductor  
laser, e.g., at 1.48  $\mu\text{m}$ .

For the evaluation of the spectroscopic parameters of  
the 1.53  $\mu\text{m}$  emission band and their composition depen-  
dence the samples were pumped, however, by a 0.8  $\mu\text{m}$   
laser to avoid any influence of the pumping light on  
the measured spectra. The composition dependence of  
the emission spectra is shown in Fig. 2. Low  $\text{Bi}_2\text{O}_3$  con-  
tent favors comparably broad and flat spectra, increas-  
ing  $\text{Bi}_2\text{O}_3$  content leads to narrower, more structured  
bands and, furthermore, to a slight red shift of the  
bands. The slight but distinct change of the lumines-  
cence band shape between 0.35 and 0.45 compositions  
might be attributed to a change in the glass microstruc-  
ture. Indications for such a change to occur at a compo-  
sition of approximately 0.40 are found in differential  
thermal analysis heating experiments carried out by  
Becker [23] on bismuth borate glasses in this composi-  
tion range.

The overall shapes of our luminescence bands differ  
from those measured recently by Chen et al. [24]. This  
might be due to a different spectral response of the  
detectors used in the measurements. As for our measure-  
ments an InGaAs photodiode with flat response in the  
spectral region near 1.5  $\mu\text{m}$  was used, the shapes seem  
to be reliable. This is further confirmed by the calcula-  
tion of the band shapes from the absorption bands using  
McCumber's theory [33] as shown in the subsequent  
section.

As characterizing spectroscopic parameters the typical  
(effective) width of the bands, defined as

$$\Delta\lambda = \int I(\lambda) d\lambda / I_{\text{max}} \quad (4)$$

the position of the maximum  $\lambda_{\text{peak}}$ , and the center of  
mass wavelengths  $\bar{\lambda}$  of the emission bands, given by

$$\bar{\lambda} = \int \lambda I(\lambda) d\lambda / \int I(\lambda) d\lambda \quad (5)$$

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Table 2

Line width  $\Delta\lambda$ , peak wavelength  $\lambda_{\text{peak}}$ , and center of mass wavelength  $\bar{\lambda}$  for the 1.53  $\mu\text{m}$  emission band in erbium doped bismuth borate glass as a function of the composition ( $\text{Bi}_2\text{O}_3$  content)

$\text{Bi}_2\text{O}_3$ content	$\Delta\lambda$ ( $\mu\text{m}$ )	$\lambda_{\text{peak}}$ ( $\mu\text{m}$ )	$\bar{\lambda}$ ( $\mu\text{m}$ )
0.25	0.0856	1.5271	1.5418
0.35	0.0824	1.5289	1.5440
0.45	0.0748	1.5301	1.5425
0.55	0.0712	1.5311	1.5429
0.65	0.0758	1.5319	1.5463

212 were calculated from the measured emission spectra.

213 The data are listed in Table 2.

214 The red shift, found for increasing  $\text{Bi}_2\text{O}_3$  content  
 215 both in the peak wavelength and in the center of mass  
 216 wavelength of the band, can be attributed to the nephel-  
 217 auxetic effect typical for rare earth ions in different envi-  
 218 ronments. It indicates an increase in the covalency of the  
 219 Er–O bonds with increasing  $\text{Bi}_2\text{O}_3$  content. Width and  
 220 flatness of the band—especially at low  $\text{Bi}_2\text{O}_3$  content—  
 221 underline the suitability of these glasses for amplifier  
 222 applications. The trend in the emission band width is  
 223 caused by an increasing optical basicity of the host glass  
 224 with increasing  $\text{Bi}_2\text{O}_3$  content [32].

225 The shape of the bands, furthermore, reflects the  
 226 trends in the Judd–Ofelt parameters. While the narrow  
 227 peak at the low wavelength side of the emission band  
 228 is due to the magnetic dipole transition—allowed and  
 229 thus independent of the ligand field, the broader struc-  
 230 ture in the band is governed by electric dipole transi-  
 231 tions—forbidden and thus dependent from the ligand  
 232 field. According to Weber [7] the line strength  $S^{\text{ed}}$  for  
 233 this band is obtained from the Judd–Ofelt parameters by

$$235 S^{\text{ed}}[{}^4\text{I}_{13/2}; {}^4\text{I}_{15/2}] = 0.019\Omega_2 + 0.118\Omega_4 + 1.462\Omega_6. \quad (6)$$

236  $\Omega_6$  is the dominating factor there-in, thus the trend in  $\Omega_6$   
 237 (see Table 1) correlates to the trend in the spectrum  
 238 shape.

### 239 3.3. Radiative efficiencies

240 Besides line width and shape, for amplifier applica-  
 241 tions the stimulated emission cross-section of a transi-  
 242 tion and its radiative quantum efficiency are of  
 243 fundamental importance. Stimulated emission cross-sec-  
 244 tions and the spectral shape of the emission bands can  
 245 be calculated from the corresponding absorption bands  
 246 by applying McCumber's theory [33]. A typical result is  
 247 shown in Fig. 3, measured absorption and calculated  
 248 stimulated emission cross-sections are plotted for the  
 249  ${}^4\text{I}_{13/2} \leftrightarrow {}^4\text{I}_{15/2}$  transition in a glass with 0.25  $\text{Bi}_2\text{O}_3$  con-  
 250 tent. To get an estimate for the accuracy of the spectral  
 251 shape calculation, the measured emission band is plot-  
 252 ted, too. From the comparison, an accuracy of about  
 253 5% for the McCumber calculation can be estimated.

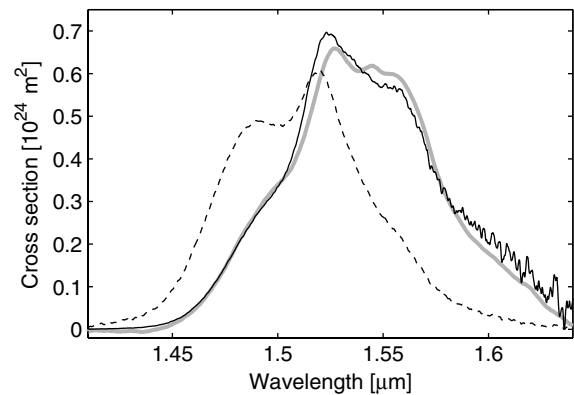


Fig. 3.  ${}^4\text{I}_{13/2} \rightarrow {}^4\text{I}_{15/2}$  absorption band of  $\text{Er}^{3+}$  in a glass with 0.25  $\text{Bi}_2\text{O}_3$  content (dashed). Applying McCumber's theory, the corresponding stimulated emission cross-section is calculated (solid black). The gray curve represents the measured emission spectrum of the  ${}^4\text{I}_{13/2} \rightarrow {}^4\text{I}_{15/2}$  transition.

254 In another way to derive the stimulated emission  
 255 cross-sections the spontaneous emission probabilities  
 256 can be used which are calculated as sum of electric  
 257 and magnetic dipole contributions from the respective  
 258 line strengths  $S^{\text{ed}}$  and  $S^{\text{md}}$  [34]

$$259 A_{J'J} = A_{J'J}^{\text{ed}} + A_{J'J}^{\text{md}} \\ = \frac{16\pi^3 e^2}{3\epsilon_0 h (2J' + 1) \lambda^3} \left\{ \frac{n(n^2 + 2)^2}{9} S_{J'J}^{\text{ed}} + n^3 S_{J'J}^{\text{md}} \right\} \quad (7)$$

261 for the transition between excited state  $J'$  with multiplicity  
 262  $(2J' + 1)$  and ground state  $J$ . The line strengths  $S$  can  
 263 be calculated according to Eqs. (1) and (2) using the  
 264 Judd–Ofelt parameters determined for the system.

265 The stimulated emission cross-section  $\sigma_e$  for the tran-  
 266 sition  $J' \rightarrow J$  then is given by [35]

$$267 \sigma_e[J'; J] = \frac{\bar{\lambda}^4 A_{J'J}}{8\pi c n^2 \Delta\lambda}. \quad (8)$$

270 Using both methods described, we calculated the stimu-  
 271 lated emission cross-sections for the  ${}^4\text{I}_{13/2} \rightarrow {}^4\text{I}_{15/2}$  tran-  
 272 sition. The values are summarized in Table 3, column I  
 273 gives the maximum values of the McCumber bands, col-  
 274 umn II the results of the derivation from the line

Table 3

Stimulated emission cross-section for the  ${}^4\text{I}_{13/2} \rightarrow {}^4\text{I}_{15/2}$  transition of  $\text{Er}^{3+}$  in bismuth borate glasses of different compositions

$\text{Bi}_2\text{O}_3$ content	Stimulated emission cross-section ( $10^{-24} \text{m}^2$ )			
	I	II	III	IV
0.25	0.70	0.66	0.72	0.67
0.35	0.77	0.62	0.84	0.70
0.45	0.83	0.79	0.94	0.71
0.55	0.86	0.78	–	–
0.65	0.82	0.67	–	–

I: McCumber calculation, II: derived from the line strengths, III: Ref. [24]—0.1%  $\text{Er}_2\text{O}_3$ , IV: Ref. [24]—0.5%  $\text{Er}_2\text{O}_3$ .



275 strengths. For comparison, data published by Chen  
276 et al. [24] for two different erbium concentrations are gi-  
277 ven in columns III and IV.

278 From the spontaneous emission probability  $A$  (Eq.  
279 (7)) the radiative lifetime  $\tau_r = A^{-1}$  for a transition can  
280 be derived. A comparison with the total lifetime  $\tau_t$  yields  
281 the radiative quantum efficiency

$$283 \eta = \tau_t / \tau_r. \quad (9)$$

284 The total lifetime can be determined experimentally by  
285 measuring the luminescence decay for the respective  
286 transitions. Typical measured decay curves are shown  
287 in Fig. 4. The linear functional relationship in the  
288 semi-logarithmic plot proves a strictly exponential  
289 behavior which can be described by a single time constant  
290  $\tau_t$  unique for each of the decay curves.

291 Decay curves have been measured for the  ${}^4I_{11/2} \rightarrow$   
292  ${}^4I_{15/2}$  transition at 0.97  $\mu\text{m}$  with 0.8  $\mu\text{m}$  excitation and  
293 for the  ${}^4I_{13/2} \rightarrow {}^4I_{15/2}$  transition at 1.53  $\mu\text{m}$  with  
294 1.48  $\mu\text{m}$  excitation for all compositions. The results—  
295 calculated radiative lifetimes  $\tau_r$ , experimentally deter-  
296 mined total lifetimes  $\tau_t$ , and radiative quantum efficien-  
297 cies  $\eta$ —are summarized in Table 4.

298 The results show that the radiative lifetimes are  
299 slightly decreasing with increasing  $\text{Bi}_2\text{O}_3$  content  
300 whereas the total lifetimes increase. This leads to an in-  
301 crease in the radiative quantum efficiencies with increas-  
302 ing  $\text{Bi}_2\text{O}_3$  content up to nearly 40% for the 1.53  $\mu\text{m}$   
303 luminescence in the glass with the highest  $\text{Bi}_2\text{O}_3$  content.  
304 Chen et al. [24] measured the decay times for composi-  
305 tions between 0.25 and 0.45 for the 1.53  $\mu\text{m}$  band. They  
306 found slightly lower values which might be due to the  
307 different excitation mechanisms (they used an excitation  
308 wavelength of 0.97  $\mu\text{m}$  whereas we used 1.48  $\mu\text{m}$ ).

309 The trend in the radiative efficiencies can be explained  
310 by the nonradiative transition probabilities. One of the  
311 most dominant of these transition processes for rare-  
312 earth ions in glasses with high phonon energy is multi-  
313 phonon relaxation. According to Layne et al. [36] the

Table 4

Radiative lifetimes, total lifetimes, and radiative efficiencies for the  ${}^4I_{11/2} \rightarrow {}^4I_{15/2}$  and the  ${}^4I_{13/2} \rightarrow {}^4I_{15/2}$  transitions (0.97  $\mu\text{m}$  and 1.53  $\mu\text{m}$ ) of  $\text{Er}^{3+}$  in glasses of different compositions ( $\text{Bi}_2\text{O}_3$  content)

$\text{Bi}_2\text{O}_3$ content	${}^4I_{11/2} \rightarrow {}^4I_{15/2}$ (0.97 $\mu\text{m}$ )			${}^4I_{13/2} \rightarrow {}^4I_{15/2}$ (1.53 $\mu\text{m}$ )		
	$\tau_r$ (ms)	$\tau_t$ (ms)	$\eta$ (%)	$\tau_r$ (ms)	$\tau_t$ (ms)	$\eta$ (%)
0.25	3.36	0.61	18.2	3.98	0.52	13.1
0.35	3.44	0.65	18.5	3.92	0.68	17.5
0.45	2.60	0.66	25.3	3.03	0.89	29.5
0.55	2.62	0.69	26.3	2.97	1.03	34.6
0.65	2.88	0.71	24.8	3.05	1.13	37.2

probabilities for multi-phonon processes depend expo- 314  
315 nentially on the phonon energies.  $\text{B}_2\text{O}_3$  host structures  
316 contribute high phonon energies,  $\text{Bi}_2\text{O}_3$  structures con-  
317 siderably lower ones. Thus in  $\text{B}_2\text{O}_3$  rich glasses high  
318 nonradiative transition probabilities, i.e., low radiative  
319 efficiencies, are found, whereas in the  $\text{Bi}_2\text{O}_3$  rich glasses  
320 the nonradiative transition probabilities decrease, en-  
321 abling higher radiative efficiencies.

### 3.4. Upconversion luminescence 322

323 Previous investigations were unable to detect upcon-  
324 version luminescence in  $\text{Er}^{3+}$  doped bismuth borate  
325 glasses, several reasons were discussed [24]. Using  
326 0.8  $\mu\text{m}$  laser excitation combined with sensitive pho-  
327 ton-counting detection techniques and prolonged inte-  
328 gration times, we at last succeeded in demonstrating  
329 such a luminescence band at 0.54  $\mu\text{m}$ , yet at fairly low  
330 intensity. The shape of the luminescence band is plotted  
331 in Fig. 5 together with its pump-intensity dependence.  
332 The slope of 1.8 found indicates a two-step excitation  
333 process for the upconversion luminescence.

334 In order to find the explanation for the unusually low  
335 intensities we calculated the radiative lifetimes for the  
336 transition  ${}^4S_{3/2} \rightarrow {}^4I_{15/2}$  responsible for the measured  
337 upconversion luminescence at 0.54  $\mu\text{m}$ . The values vary

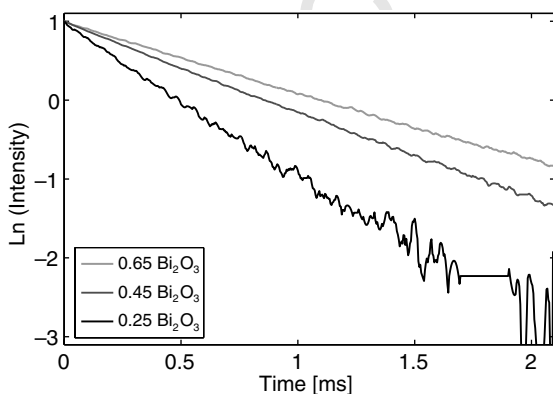


Fig. 4. Luminescence decay for the  ${}^4I_{13/2} \rightarrow {}^4I_{15/2}$  transition of  $\text{Er}^{3+}$  in bismuth borate glasses of different compositions.

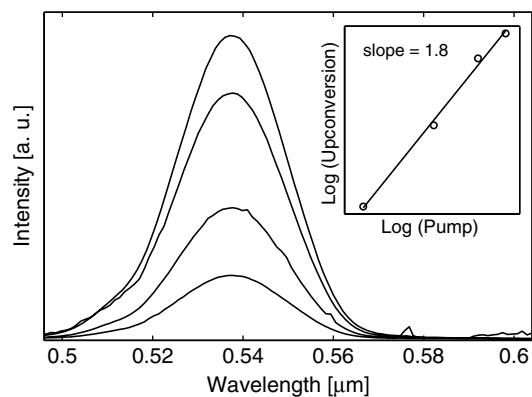


Fig. 5. Upconversion luminescence in  $\text{Er}^{3+}$  doped bismuth borate glass ( $\text{Bi}_2\text{O}_3$  content 0.25). In the inset the dependence of the upconversion intensity on pump intensity is sketched.

338 from 0.3 ms to 0.25 ms for the  $\text{Bi}_2\text{O}_3$  content varying  
 339 from 0.25 to 0.65. In contrast to this, the measured de-  
 340 cay times for this luminescence band (when excited with  
 341 pulsed 0.53  $\mu\text{m}$  laser light) lie at approximately 100 ns  
 342 [36]. Thus the radiative quantum efficiencies for this  
 343 transition are in the order of only 0.03%. This extremely  
 344 low value is due to a high nonradiative transition rate  
 345 from the  $^4\text{S}_{3/2}$  level to the  $^4\text{F}_{9/2}$  level of  $\text{Er}^{3+}$  caused by  
 346 the high phonon energies of the  $\text{B}_2\text{O}_3$  host structures  
 347 [36].

### 348 3.5. Higher doping

349 To study the effect of higher doping, absorption and  
 350 emission spectra for samples with tripled and 10-fold er-  
 351 bium content have been measured. For the concentra-  
 352 tion range investigated, all absorption bands scaled  
 353 linearly with the erbium concentration. That means un-  
 354 changed Judd–Ofelt parameters.

355 The measured luminescence decay times, however,  
 356 showed a slight decrease—from 0.61 ms to 0.58 ms for  
 357 the 0.97  $\mu\text{m}$  band and from 0.52 ms to 0.45 ms for the  
 358 1.53  $\mu\text{m}$  band—for the erbium content increasing by a  
 359 factor of ten. This corresponds to a decrease in the radi-  
 360 ative quantum efficiencies from 18.2% to 17.1% and  
 361 from 13.1% to 11.4%, respectively.

### 362 4. Conclusions

363 Optical properties—including refractive indices,  
 364 absorption and emission spectra, decay times of selected  
 365 transitions, and upconversion luminescence—of erbium  
 366 doped bismuth borate glasses have been measured for a  
 367 wide composition range. Varying the  $\text{Bi}_2\text{O}_3$  content  
 368 from 0.25 to 0.65, the refractive index of the glasses  
 369 can be tuned between 1.8 and 2.3. For the description  
 370 of the refractive indices in the wavelength range from  
 371 0.4 to 1.5  $\mu\text{m}$  and in the full composition range a gener-  
 372 alized Sellmeier formula is provided. From the absorp-  
 373 tion spectra, the Judd–Ofelt parameters are derived for  
 374 all glass compositions. They can be used to calculate  
 375 oscillator strengths, line strengths, transition probabili-  
 376 ties, radiative lifetimes, and stimulated emission cross-  
 377 sections—applying the usual formulas.

378 From the comparison between the calculated radi-  
 379 ative lifetimes and the measured luminescence decay times  
 380 the radiative quantum efficiencies for the  $^4\text{I}_{11/2} \rightarrow ^4\text{I}_{15/2}$   
 381 and the  $^4\text{I}_{13/2} \rightarrow ^4\text{I}_{15/2}$  transitions (0.97  $\mu\text{m}$  and 1.53  
 382  $\mu\text{m}$ , respectively) were derived. The maximum quantum  
 383 efficiency of nearly 40% was found for the 1.53  $\mu\text{m}$  tran-  
 384 sition in the glass with a  $\text{Bi}_2\text{O}_3$  content of 0.65. The max-  
 385 imum linewidth of this transition, 0.093  $\mu\text{m}$ , however,  
 386 was obtained for a bismuth content of 0.25. Stimulated  
 387 emission cross-sections for the  $^4\text{I}_{13/2} \rightarrow ^4\text{I}_{15/2}$  transition

vary from about  $0.75 \times 10^{-24} \text{ m}^2$  for low to about  
 $0.60 \times 10^{-24} \text{ m}^2$  for high  $\text{Bi}_2\text{O}_3$  content.

Upconversion luminescence at 0.54  $\mu\text{m}$  could be dem-  
 onstrated, however with extremely low intensity. The  
 reasons for this are nonradiative multi-phonon transi-  
 tion processes, depleting the  $^4\text{S}_{3/2}$  level, which are highly  
 efficient in borate glasses due to the high phonon ener-  
 gies of the borate host structures.

Erbium doped bismuth borate glasses have thus been  
 proven to be interesting materials for large range of  
 optical applications like fiber amplifiers or lasers. Vary-  
 ing the composition of the host glass, the application-  
 relevant parameters can be widely adjusted to suit the  
 needs of the particular application.

### Acknowledgments

The work was supported by the Federal State of Nie-  
 dersachsen within the Graduate School ‘Synthesis and  
 Characterization of Surfaces and Interfaces Assembled  
 from Clusters and Molecules’.

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